

Note

Bis(2-acetamido-3,4,6-tri-*O*-acetyl-2-deoxy- β -D-glucopyranosyl)amine obtained by a fusion reaction

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Under fusion conditions, glycosides can be formed by the elimination of acetic acid¹, trimethylsilyl halide², or trihalogenoacetic acid³ from acylated sugars and aglycons.

We now report a novel fusion reaction, namely, the conversion of 2-acetamido-3,4,6-tri-*O*-acetyl-2-deoxy- β -D-glucopyranosylamine (**1**) into bis(2-acetamido-3,4,6-tri-*O*-acetyl-2-deoxy- β -D-glucopyranosyl)amine (**2**) by the elimination of ammonia. Compound **2**, which was obtained crystalline (57%), was originally isolated⁴ as a by-product in the preparation of **1**.

When the crude fusion product was eluted from Sephadex LH-20 with ethanol, **2** was obtained first, followed by maltose octa-acetate (reference added), and then 2-acetamido-1,3,4,6-tetra-*O*-acetyl-2-deoxy-D-glucose. The n.m.r. spectrum of **2**

TABLE I

DATA ON TWO BIS(PER-*O*-ACETYLGLYCOSYL)AMINES

Compound ^a	Yield M.p. (%)		[α] _D ²⁰ (c, chloroform)	R _F ^b	Found (%)		
					C	H	N
Bis(2,3,4,6-tetra- <i>O</i> -acetyl-D-galactopyranosyl)amine ^c	47	Amorphous	+59° (1.2)	0.72	49.64	5.84	1.84
Bis(2,3,4,6-tetra- <i>O</i> -acetyl- α -D-glucopyranosyl)amine ^d	18	222–224°	+86° (0.64)	0.70	49.59	5.81	2.00

^aCalc. for C₂₈H₃₉NO₁₈: C, 49.63; H, 5.80; N, 2.07. ^bT.l.c., benzene–ethyl acetate (1:1). ^cAnomeric configuration not established; $\nu_{\text{max}}^{\text{KBr}}$ 3400 (NH), 1750, 1240 (C=O, C–O of OAc), and 1090–1010 cm^{−1} (C–O). N.m.r. data (CDCl₃): δ 5.32 (2 H), 5.2–4.9 (5 H), 4.3–3.7 (7 H), 3.00 (1 H, NH), 2.13, 2.00, 1.98, and 1.93 (24 H, 8 OAc). ^d $\nu_{\text{max}}^{\text{KBr}}$ 3400 (NH), 1750, 1250 (C=O, C–O of OAc), and 1100–1000 cm^{−1} (C–O). N.m.r. data (CDCl₃): δ 5.4–3.9 (14 H), 3.00 (1 H, NH), and 2.07–1.99 (24 H, 8 OAc); lit.⁷: α anomer, m.p. 216–217°, [α]_D +87° (chloroform); β anomer, m.p. 190–192°, [α]_D +7.6° (chloroform).

indicated the presence of three protons (NH) exchangeable with deuterium, and a ratio of intensities of acetyl and methine plus methylene protons of $\sim 12:7$ after the deuterium exchange. Compound **2** had i.r. bands for NAc and OAc groups, and the electron-impact and chemical-ionisation (NH_3) mass spectra contained signals for M^+ at m/e 675 and MH^+ at m/e 676, respectively.

Compound **2** was also obtained when zinc chloride was used as catalyst, but considerable degradation occurred when toluene-*p*-sulphonic acid was employed in this capacity.

Two other diglycosylamines were also prepared by the novel fusion procedure (Table I) from the corresponding glycosylamines.

EXPERIMENTAL

Mass spectra were recorded with a Shimadzu LKB-9000 mass spectrometer at an accelerating potential of 3,500 V, and a source temperature of 250° for electron impact (60 eV) and 230° for chemical ionisation (NH_3 , 500 eV). The other methods have been described previously⁶.

Bis(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyl)amine (2). — Compound **1** {228 mg, m.p. $156\text{--}157^\circ$, $[\alpha]_D^{20} -22^\circ$ (*c* 0.5, chloroform)} was fused at $\sim 160^\circ$ for 5–10 min with the exclusion of moisture and under weak suction. A solution of the resulting, slightly brown syrup in ethanol (~ 20 ml) was decolorised with activated charcoal and concentrated *in vacuo* to give a syrup (207 mg, 93%) which crystallised from ethanol, to yield **2** (126 mg, 57%), m.p. and mixture m.p. $259\text{--}262^\circ$ (dec.), $[\alpha]_D^{20} -27^\circ$ (*c* 1.3, chloroform) {lit.⁵ m.p. 260° , $[\alpha]_D -32^\circ$ (*c* 2, chloroform); cf. m.p. $245\text{--}246^\circ$, $[\alpha]_D +44^\circ$ (*c* 2, chloroform) for the $\alpha\beta$ compounds⁵}; $\nu_{\text{max}}^{\text{KBr}}$ 3360 (NH), 1750 and 1250 (C=O and C–O of OAc), 1660 and 1540 (C=O and NH of NHAc), and $\sim 1050\text{ cm}^{-1}$ (C–O). N.m.r. data (CDCl_3): δ 6.05 (d, 2 H, NH), 5.4–3.4 (14 H, -CH- and -CH₂), 2.15 (1 H, NH), 2.10, 2.00, and 1.95 (3 s, 15 H, 2 NAc, 5 OAc). Mass spectrum: m/e 675 (M^+), 616 ($\text{M} - \text{OAc}$), 556 ($616 - \text{AcOH}$), 496 ($556 - \text{AcOH}$), 436 ($496 - \text{AcOH}$), 330, 270 ($330 - \text{AcOH}$), 210 ($270 - \text{AcOH}$), and 150 ($210 - \text{AcOH}$). The ion m/e 330 is identical with that given by 2-acetamido-1,3,4,6-tetra-O-acetyl-2-deoxy- α -D-glucopyranose⁸.

Anal. Calc. for $\text{C}_{28}\text{H}_{41}\text{N}_3\text{O}_{16}$: C, 49.78; H, 6.12; N, 6.22. Found: C, 49.69; H, 6.08; N, 6.17.

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REFERENCES

- 1 B. HELFERICH AND E. SCHMITZ-HILLEBRECHT, *Ber.*, 66B (1933) 378–383.
- 2 T. NISHIMURA AND I. IWAI, *Chem. Pharm. Bull.*, 12 (1964) 352–356.

- 3 K. ONODERA AND H. FUKUMI, *Agric. Biol. Chem.*, 27 (1963) 526-529.
- 4 A. YAMAMOTO, C. MIYASHITA, AND H. TSUKAMOTO, *Chem. Pharm. Bull.*, 13 (1965) 1036-1041.
- 5 C. H. BOLTON, L. HOUGH, AND M. Y. KHAN, *Biochem. J.*, 101 (1966) 184-190.
- 6 S. HIRANO AND W. OHASHI, *Carbohydr. Res.*, 59 (1977) 285-288.
- 7 P. BRIGL AND H. KEPPLER, *Hoppe-Seyler's Z. Physiol. Chem.*, 180 (1929) 38-63.
- 8 R. C. DOUGHERTY, D. HORTON, K. D. PHILIPS, AND J. D. WANDER, *Org. Mass Spectrom.*, 7 (1973) 805-816.